

## The use of hybrid aeration system in wastewater treatment

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### Abstract

The objective of this study was to demonstrate and assess the effectiveness of hybrid aeration system and to identify how equipment improvements and operational adjustments will impact the efficiency of the aeration process. Evaluating of oxygen transfer rate and oxygenation capacity were the functions to assess such system's efficiency. Periodical runs, with enhancement of two laboratory glass basin model with (58) liter for each were conducted, one was for water and the other for wastewater where D.O analysis were taken stimulatingly at peered time intervals. Speed mixing and air discharge were the main parameters that used in detecting system efficiency. The study showed that the best result was obtained at air discharge about ( 1490 ml /min. ) and mixer (50 RPM), that alpha factor and oxygenation capacity were (1.12) and (43.2 mg O<sub>2</sub>/L . min.), respectively. However, the study concluded that increasing mixing speed and air discharge may affect conversely on bubble size and consequently on oxygen transfer rate in aeration systems of wastewater treatment plant.

**Keywords:** Oxygen transfer, Hybrid Aeration, Alpha factor, Aeration Systems.

### Introduction

The requirements to improve effluent quality have led to the establishment of better design procedures for activated-sludge plants. Aeration process in most wastewater treatment plant consumes a major portion of plant's power, thus, the aeration systems represent the impact factor in designing an economical and efficient operated plant (Smith, 2001).

Although there are many types of aeration systems, the two basic methods of aerating wastewater are through mechanical surface aeration to entrain air into the wastewater by agitation, or by introducing air or pure oxygen with submerged diffusers. Fine-bubble diffused-air techniques are potentially more efficient than mechanical surface-aeration systems, but low oxygen-transfer efficiency is often observed in practice under conditions of high aeration tank solids loading. Conversely, surface-aeration systems appear to be able to operate at reasonable aeration efficiencies under conditions of high loading, but they are not as effective as diffused-air systems in low-rate processes (Chambers *et al.*, 1998).

The focus of this paper was to assess a simple baseline model of a municipal wastewater treatment of aeration system's facilities, in order to understand the various

operational strategies and equipments selection in wastewater treatment. As part of the calculation process of comparing aeration system equipments, the relative rate of oxygen transfer in wastewater compared to clean water must be established (alpha value).

| Aeration System             | Typical Alpha ( $\alpha$ ) |
|-----------------------------|----------------------------|
| Course Bubble Diffusers     | 0.80                       |
| Fine Bubble Diffusers       | 0.45                       |
| Jet Aeration                | 0.75                       |
| Surface Mechanical Aerators | 0.85                       |
| Submerged Turbines          | 0.85                       |

**Table. (1):** Typical Alpha Values (USEPA, 1989).

Another important function of the aeration equipment is to provide adequate mixing in the tanks to prevent solids from settling. It is typical for many municipal wastewater treatment facilities to have BOD load to be the predominant factor for the aeration system to satisfy during the day, but when BOD loads decrease during the late evening hours, adequate mixing in the tank may be the controlling energy requirement. Table 2. Shows typical minimum mixing values for aeration tanks (USEPA, 1989; Metcalf and Eddy, 2003).

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| Type of Aeration System         | Mixing Requirement         |
|---------------------------------|----------------------------|
| Course Bubble Diffused Aeration | 20 to 30 scfm/1000 cu.ft.  |
| Fine Bubble Diffused Aeration   | 7 to 10 scfm / 1000 cu.ft. |
| Mechanical Surface Aeration     | 0.6 to 1.15 hp/1000 cu.ft. |

**Table. (2):** Typical Aeration Tank Mixing Requirement.

For most aeration designs alpha factor used in many fine bubble activated sludge systems is as low as 0.5; this results in a process aeration efficiency (PAE) that is already reduced to half of the clean-water efficiency (Smith, 2001).

EPA conducted a study depending on the place of using aeration equipment in plug flow reactor. The study comes out with that when using fine bubble aeration alpha value was 0.3 at reactor influent while it was 0.8 at effluent.

In survey study conducted by (Chambers *et al.*, 1998). the activated-sludge plants using hybrid aeration systems at Whitlingham, Blackburn Meadows and Wanlip produce consistently good effluent quality (measured in terms of BOD, SS and NH<sub>3</sub>). The aeration efficiency of the plants varies between 1.0 and 1.4 kg/kWh and depends upon the assumptions used in design being reflected in actual sewage characteristics and operating conditions. The installation of FBDA (Fine Bubble Diffused Aeration) equipment in relatively low-loaded regions of the aeration tanks does not result in diffuser sliming or blockage with consequent performance deterioration.

A model for oxygen transfer rate in diffused air systems was proposed by (Bayramoglu *et al.*, 2000). The model parameters were found by means of non linear regression analysis, using plant data given in the literature. The proposed model may be useful for the preliminary design of diffused air aeration tanks.

In order to optimize aeration in the activated sludge processes, an experimentally validated numerical tool was proposed by (Yannick, 2007). based on computational fluid dynamics and able to predict flow and oxygen transfer characteristics in aeration tanks equipped with fine bubble diffusers and axial slow

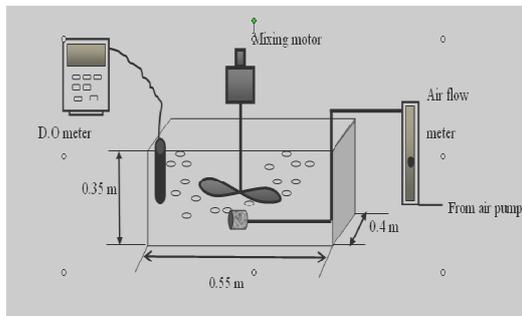
speed mixers. For four different aeration tanks (1; 1493; 8191 and 29.313m<sup>3</sup>). Predicted oxygen transfer coefficients were within ±5% of experimental results that ranged from (2.19× 10<sup>3</sup> – 1.71× 10<sup>3</sup> s<sup>-1</sup>) for different operating conditions (varying pumping flow rates of the mixers and air flow rates).

The effect of airflow rate and the level of diffusers submergence, on the oxygen transfer rate of diffused air systems was determined by (Al-Ahmady, 2006). He derived an individual mathematical models to describe the effect of each parameter. The results of the study showed that, increasing the airflow rate at fixed water depth and diffusers submergence enlarge the oxygenation capacity of the system. The equation, which controls this relationship, is linear. At diffusers submergence of 4.6 m, the slope of the equation was 11.8. With reduce the depth of diffusers to about 0.4 m; the slope of equation was decrease to 2.3. At constant airflow rate, the depth of diffusers has a significant effect on both of the oxygenation capacity and the oxygen transfer efficiency of the system. Exponential form of equation is shown to be efficient in expressing the relationship between the submergence and the oxygenation capacity. At 0.4 m, diffusers submergence, the oxygen transfer efficiency was 1.8 whereas; this value is increased to about 11.5 at 4.6m submergence.

(Zhen He *et al.*, 2003). measures the oxygen transfer capacity in clean water by using desorption and absorption techniques. For the absorption method, the mean value of KLa<sub>20</sub> was obtained as 8.60 h<sup>-1</sup> with a water volume of 14L. Meanwhile, standard oxygen transfer efficiency (SOTE) was shown in the range of 4.5-4.9% with water depth 0.3 m by correcting the airflow condition. Desorption measurement was investigated to certify the influence of water depth on SOTE. Five different water depths (0.24, 0.26, 0.27, 0.30, and 0.32m) are chosen as test objective. Due to the small interval of water depths chosen in this experiment, it was difficult to measure the exact change of SOTE with water depth. However, SOTE increased with the water depth.

**Experimental Work**

According to (Pre-European Standards, 1999). Oxygen transfer information was obtained by using laboratory (bench-scale) apparatus. All tests reported herein were carried out by using two equally laboratory scale tanks (0.55m × 0.40 m × 0.35m), as shown in figure (1), which were operated in the batch mode. The DO probe was installed at half water depth for absorption measurement. Within the tank, the diffusers were placed 10 cm from the tank’s bottom. Prior to each group of runs, the tanks were drained, cleaned and filled with clean water to avoid salt accumulation. The experiment program was divided into four groups. In each group, the effect of air discharge and mixing intensity, on the studied parameters were investigated as shown in table (3).



**Fig. 1:** Configuration of the experimental work.

Table (3) below shows the experiment design accordingly to the experiment specification group which coincided mixing intensity and air flowrate in both water and wastewater basins model.

| Sample type | Group No. | Air discharge (m <sup>3</sup> / day) | Mixing intensity (RPM) |
|-------------|-----------|--------------------------------------|------------------------|
| Tap water   | 1         | 5.35                                 | 50                     |
|             | 2         | 5.35                                 | 100                    |
|             | 3         | 10.7                                 | 50                     |
|             | 4         | 10.7                                 | 100                    |
| Wastewater  | 1         | 5.35                                 | 50                     |
|             | 2         | 5.35                                 | 100                    |
|             | 3         | 10.7                                 | 50                     |
|             | 4         | 10.7                                 | 100                    |

**Table. (3):** Demonstration of Experiments Design.

The major shortage in these tests that they were carried out in small scale so, their results mainly limits by the test circumstances.

**Calibration of DO Probe**

Calibration of DO probe (EXTECH; model 407510) is necessary for the precise measurement. The methods used in this experiment consist of zero check and saturation check. The zero check was run every time after the probe was turned on, by putting probe into a 1 L cylindrical glass added by 1 g sodium sulfate and 1 mg cobalt. With excess chemical, the water in the tube should contain zero dissolved oxygen. Saturation check was done before and after measurement, according to the operation manual for the probe.

**Temperature Measurement**

The temperature of test water was read from the DO meter. The variation of temperature should be with an accuracy of ± 0.5 °C at the beginning and the end of each test (German ATV Standards ,1996).

KLaT was determined by evaluation of an oxygen transfer test in clean water and wastewater at a certain aeration setting and at a certain temperature. So, it was converted to the standard temperature of T = 20 °C as follows (Metcalf and Eddy ,2003):

$$(KLa)T = (KLa)_{20} \times (\theta)^{T-20} \dots\dots\dots (1)$$

Where:

(KLa)<sub>20</sub> = value of KLa at 20°C.

(KLa)<sub>T</sub> = value of KLa at test water temperature.

T = test water temperature, °C.

θ = 1.024

**Airflow Rate Measurement**

A laboratory flow meter was used to measure the airflow rate. The accuracy of the flow meter has been checked before application. The inverse measuring cylinder filled with water was used to calibrate air flowrate. The airflow of 1490 ml/min and 2980 ml /min. was applied in the measurement.

The oxygen transfer tests were conducted in parallel using tap water and domestic wastewater for comparative purposes and deducing the correction factor of Oxygen transfer coefficient. There are several methods of experimental determination of mass transfer coefficients. The so-called clean water non-steady state method was selected in this study. The unsteady state test

is presently the most broadly accepted test procedure. This method was practiced according to the procedure advised by (Eckenfelder, 2000; Ramalho, 1977). which involves.:

1. Removing the dissolved oxygen in the aeration tank by adding sodium sulfate  $\text{Na}_2\text{SO}_3$  (8 mg/l per 1mg/l of dissolved oxygen) and cobalt chloride ( $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ) concentration of 0.05 mg/l to provide a minimum  $\text{Co}^{+2}$  of 1.5 mg/l.
2. Thoroughly mix the tank contents.
3. Recording of the oxygen concentration increase while aerating the previously deoxygenated water.
4. Recording the temperature and measuring oxygen saturation value from tables (Metcalf and Eddy, 2003).
5. Computing the oxygen transfer rate in accordance to Eq. (2) as shown below.

It makes no difference to aerate with air or with oxygen because the value of oxygen transfer coefficient (KL) is only dependent on the hydrodynamic properties of the system (Metcalf & Eddy, 2003). Relationships that describe the rate of water oxygenating in step (3), and formula for computing the oxygen mass transfer coefficient are presented in the following:

$$\frac{dc}{dt} = KLa (C_{sw} - CL) \dots\dots\dots (2)$$

Where:

$dc/dt$ : the rate of change of the oxygen concentration

$KLa$ : oxygen transfer rate across the gas liquid film, (1/min.)

$CL$ : oxygen concentration at time (t), (mg/L)

$C_{sw}$ : the saturation concentration of oxygen in water and wastewater, (mg/L)

t: the time, (min.)

The representation of eq. (2) was depicted in fig (2), showing the determination of  $KLa$ .

The difference ( $C_{sw} - CL$ ) between saturation value and actual concentration of oxygen ( $CL$ ) in the body of the liquid phase is usually called oxygen deficit. The oxygen transfer rate was determined by integrating of this equation.

From equation (2), the initial oxygen uptake rate at  $CL=0$ , is

$$\frac{dc}{dt} = OC = KLa (C_{sw}) \dots\dots\dots (3)$$

Where:

$OC$  = the oxygen transfer capacity of the system, ( $\text{grO}_2/\text{m}^3\text{water} \cdot \text{hr}$ )

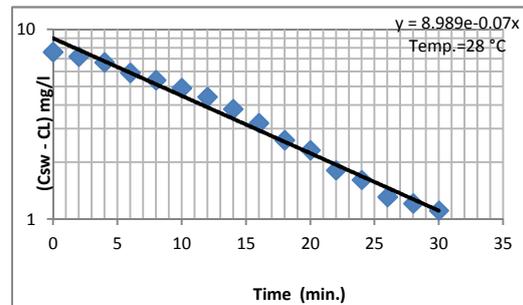
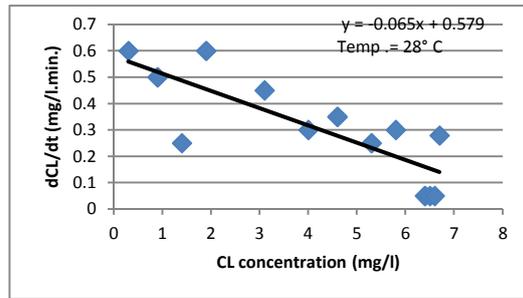
To introduce the effect of wastewater on the results obtained from this procedure, the following equation for admixtures correction factor is usually used (Kalinske *et al.*, 1973).

$$\alpha = \frac{KLa_{wastewater}}{KLa_{water}} \dots\dots\dots (4)$$

Where:

$KLa$  = Oxygen transfer coefficient for the specific admixture (1/hr).

$\alpha$  = the admixtures correction factor.



**Fig. 2:** Sample of the experimental results depicting oxygen transfer coefficient determination for (A-water and B-wastewater).

In table (4), the values of correction factor for some types of wastewater are represented.

| Type of Wastewater                   | $\alpha$ – factor |
|--------------------------------------|-------------------|
| Domestic wastewater                  | 0.85              |
| Dairy wastewater                     | 0.82              |
| Wastewater from chemicals industries | 0.7               |
| Pulp and paper industry wastewater   | 0.7               |
| Wastewater from acidic industries    | 0.5               |

**Table. ( 4):** Values of The Correction Factor for Some Types of Wastewater (Khudenko and Shpirt, 1986).

**Results and Discussion:**

Fig. (3). depicted the effect of mixing intensity and air flowrate on oxygen transfer rate ( $KLa$ ) for both water and wastewater samples. Obviously, mixing intensity and air flowrate (group No.2 and 4) have noticeable

effect on such factor that increasing mixing intensity to 100 RPM increased (KLa of water sample) from 1.5 to 2.88 hr<sup>-1</sup> at fixed air flowrate 5.35 m<sup>3</sup>/day, while duplicating air flowrate to 10.7 m<sup>3</sup>/day increased (KLa) from 2.88 to 3.42 hr<sup>-1</sup> at 100 RPM. It seemed that mixing intensity were more effected than air flowrate on (KLa), about 48% increment. Conversely, air flowrate increment was decreased (K La) of wastewater sample from 2.1 to 1.8 hr<sup>-1</sup> at 100RPM, about 85%, (see table 5). Practically, it was shown that the presence of certain organic chemicals (surfactants) was increased resistance at the gas/liquid interface, which impedes molecular diffusion, D, and forms rigid surface on bubbles that reduced the liquid film coefficient KL and overall mass transfer coefficient, KLa, to about 55% depending on type of surfactant, concentration and type of wastewater (Shifrin *etal*,1977;Wagner and Johannes, 1986).

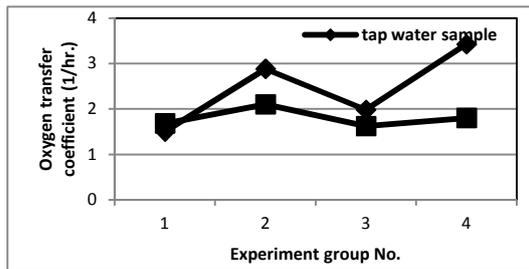


Fig. (3): Relationship between oxygen transfer rate with mixing intensity and air flowrate

Since, the major component in wastewater effecting KLa in fig (3). was surfactants which was a highly significant impact on the gas-liquid interface, decreasing the surface tension of the interface and making transfer more difficult, which increased as the bubble size gets smaller (Mark,2005) So , fine bubble diffusers have adverse effect on Oxygen correction factor ( $\alpha$ ). While, in activated sludge systems using mechanical aerators, surfactants help in producing smaller water droplets, increasing the available surface area for oxygen

transfer. Thus, the alpha factor can actually be above 1.0 (Smith, 2001). Basically, oxygen correction factor ( $\alpha$ ) depends on KLa coefficient, group (1) that had low mixing intensity of 50 RPM and air flowrate 5.35 m<sup>3</sup>/day, which produced relatively large bubble size, gave the best results for this factor about (1.12) as depicted in fig (4).

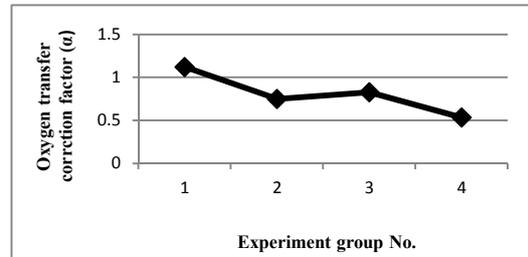


Fig. (4): Relationship between oxygen transfer correction factor with mixing intensity and air flowrate

Furthermore, fig (5) demonstrates the effect of studied parameter on oxygenation capacity (OC). Since oxygenation capacity was related to oxygen transfer coefficient, its obviously that group (1) was the dominating factor for this parameter. Oxygenation capacity was 43.22 mg O<sub>2</sub> /min for wastewater, referring to table (5) its remarkable that reducing mixing intensity increased the (OC) to about 63% and 61% at 5.35 and 10.7 m<sup>3</sup>/day of air flowrate respectively, deducing that power saving in wastewater treatment plant may be accomplished by reducing the necessity to the high mixing intensities in the operation processes of the plant.

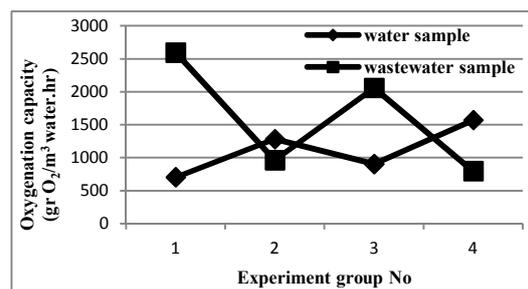


Fig. (5): Relationship between oxygenation capacity with mixing intensity and air flowrat

| Parameter                                   | Q air =5.3 m <sup>3</sup> /d<br>RPM = 50 | Q air =5.3 m <sup>3</sup> /d<br>RPM = 100 | Q air =10.7 m <sup>3</sup> /d<br>RPM = 50 | Q air =10.7 m <sup>3</sup> /d<br>RPM = 100 |
|---|--|---|---|--|
| Alpha                                       | 1.17                                     | 0.729                                     | 0.818                                     | 0.526                                      |
| OCWW<br>mg O <sub>2</sub> /l.min            | 43.224                                   | 16.039                                    | 34.296                                    | 13.261                                     |
| OCW<br>mg O <sub>2</sub> /l.min             | 11.717                                   | 21.357                                    | 15.059                                    | 26.166                                     |
| Kla <sub>WW</sub> @20°C<br>1/hr             | 1.68                                     | 2.1                                       | 1.62                                      | 1.8  |
| KLa <sub>W</sub> @20°C<br>1/hr              | 1.5                                      | 2.88                                      | 1.98                                      | 3.42                                       |
| Oxygenation<br>efficiency<br>of water%      | 31.73                                    | 57.84                                     | 20.39                                     | 62.09                                      |
| Oxygenation<br>efficiency<br>of wastewater% | 117                                      | 64.33                                     | 46.44                                     | 30.52                                      |
| Oxygen<br>absorption of<br>wastewater %     | 1.17                                     | 1.44                                      | 0.56                                      | 0.59                                       |
| Oxygen<br>absorption of<br>water %          | 1.06                                     | 1.97                                      | 0.69                                      | 1.21                                       |

Table. (5): Summary of The Results

In fig (6) the effect of sample type (wastewater, tap water) was obtained, since there was shortage in oxygen in waste sample due to degradation process the oxygen molecules was substituted rapidly. Group (1) indicated best result for oxygenation efficiency about 117 gr O<sub>2</sub>/m<sup>3</sup> air assuring the needles to the high power for both mixing intensity ,air discharge .

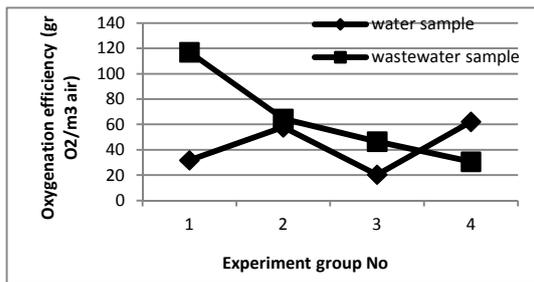


Fig. (6): Relationship between oxygenation efficiency with mixing intensity and air flowrate

For the oxygen absorption ratio related to experiment parameters in fig (7), it was unquestionable that the absorption ratio for wastewater was less than for tap water this may contributed to the resistance of wastewater sample to oxygen transmission to the bulk of the liquid by diffusion and convection.

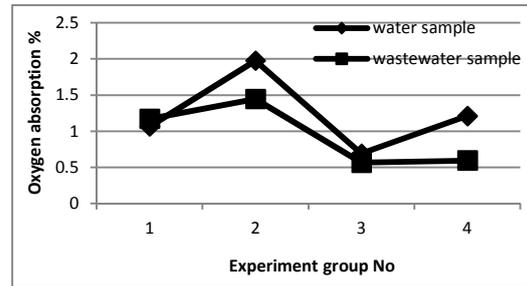


Fig. (7): Relationship between oxygen absorption with mixing intensity and air flowrate

**Conclusions:**

The use of hybrid aeration system in wastewater treatment was more efficient and economical in power saving than the conventional systems, since high values of oxygen transfer coefficients , oxygenation capacity and the correction factor of oxygen transfer were obtained at low mixing intensities and air flowrate of 50RPM and 5.35 m<sup>3</sup>/day respectively.

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## الملخص العربي

كان الهدف من هذه الدراسة هو إستعراض وتقييم فعالية نظام التهوية الهجين (المركب) ووصف آليات التشغيل المعتمدة في أنظمة التهوية ودورها في تحسين عملية التهوية من جهة وتوفير الطاقة من جهة أخرى. تم إستنباط المعاملات الخاصة بمعدل إنتقال الأوكسجين وسعة إنتقال الأوكسجين في الماء وغيرها من المعاملات في تقييم كفاءة هذا النظام. إستخدمت عدة تجارب وبشكل دوري وذلك بالاستعانة بأحواض زجاجية عدد اثنين سعة ٥٨ لتر للحوض الواحد أحدهما للماء والأخر لمياه الفضلات المدنية حيث تم إجراء الفحوصات بشكل متواز للحوضين وعند فترات زمنية متساوية. تم دراسة تأثير كل من سرعة المزج وتصريف الهواء على المعاملات المستخدمة في تقييم كفاءة النظام، حيث تبين أن تصريف الهواء / ml min1490 وسرعة مزج 50 RPM أعطيا أفضل النتائج حيث كانت قيمة معامل التصحيح للمزيج ( $\alpha$ ) وسعة إنتقال الأوكسجين مساوية لـ 1.12 و 43.2 mg O<sub>2</sub>/L.min على التوالي. تم الاستنتاج بان زيادة سرعة المزج وتصريف الهواء قد يؤثران بصورة سلبية على حجم الفقاعة وبالتالي على معدل إنتقال الأوكسجين في أنظمة التهوية في محطات معالجة مياه الفضلات.

الكلمات الدالة: إنتقال الأوكسجين، نظام التهوية الهجين، معامل  $\alpha$ ، أنظمة التهوية.